

International Journal of Research Publication and Reviews

Journal homepage: www.ijrpr.com ISSN 2582-7421

Preparative HPLC and its Applications: A Review

¹Aarti Patil, ²Saurabh Chavan

¹M Pharmacy Student, PES Modern College of Pharmacy (For Ladies), Moshi, Pune.
²M Pharmacy Student, SCES Indira College of Pharmacy, Tathawade, Pune.

ABSTRACT:

Preparative chromatography is only one of many vital techniques that can be used effectively as a precursor to determine the structures of unknown impurities or clean up the sample to meet specification. Dealing with these unknown impurities, for example in finished products, stability samples or discovered compounds, can be very challenging, as failure to identify or remove them could interrupt product supply, delay clinical trials, or disrupt with toxicology studies. Very often there is a regulatory requirement to disclose the structure of new impurities above a certain limit. Financial loss to the pharmaceutical, agrochemical, food and consumer and fine chemical industries can be huge.

Introduction:

The term preparative HPLC is usually associated with large columns and high flow rates. However, the preparative HPLC experiment does not determine the size of the instrumentation or the amount of mobile phase pumped by the system, but rather the goal of the separation. The goal of preparative HPLC is the isolation and purification of a valuable product. Since preparative HPLC is a relatively expensive technique compared to traditional purification methods such as distillation, crystallization or extraction, it has only been used for rare or expensive products. With the increasing demand to produce highly pure compounds in various quantities for activity, toxicology and pharmaceutical screenings, the scope of preparative HPLC is changing.

Objectives of Preparative HPLC:

Three important parameters used to assess the outcome of a preparative run are product purity, yield, and power. Since the parameters are interdependent, it is not possible to optimize the preparative HPLC method with respect to all three parameters.

Principle of Preparative HPLC:

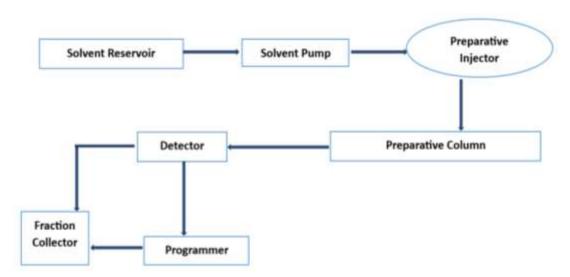
The working principle of prep. HPLC is the same as analytical HPLC, which involves separation based on adsorption. In addition to analytical HPLC components, prep. The HPLC fraction collector is located behind the detector. In prep. In HPLC, the monitored analyte is collected in the fraction collector after detection by the detector. By purification, we can reuse this sample for detection if the experiment fails under any conditions.

Comparison between analytical and preparative HPLC:

Parameter	Analytical	Preparative
Sample collection	Sample goes from detector to waste	Sample goes from detector to fraction collector
Column Size (mm)	120 - 250 x 2-4.6	120 - 250 x 20-62
Particle size (µm)	Upto 5	>10
Stationary Phase (g)	Upto 5	50-450
Flow rate (ml/min)	0.1-2	100-1000
Sample size (mg)	0.01-2	1-700
Sample concentration	High	Low

Solubility of Sample	Solubility of Sample in mobile phase usually not important	Solubility of Sample in mobile phase is very important	
Capacity of solvent reservoir	In litre	Several gallons	
Diverter Valve	Absent	Present	
Fraction Collector	Absent	Present	
Injector Capacity	Upto 100µ1	0.1-100 ml	
Pump Capacity	Low	High	
Sensitivity of detector	High	Low	
Analytical Use	For Quantitation and/or identification of compound	For isolation and/or purification of compound	

Instrumentation:



1. Solvent reservoir:

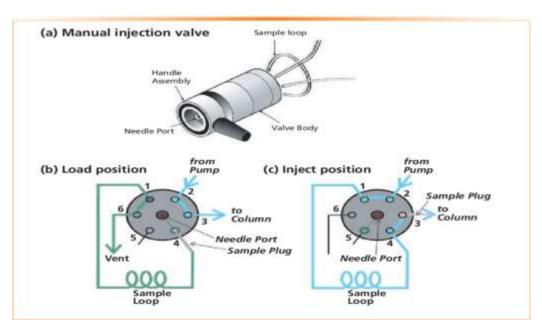
Large capacity (up to several gallons) containers made of glass or stainless steel are used in preparation. HPLC. The material of the structure varies according to the type of material. A biocompatible material is used for biologically sensitive material.

2. Pump:

Prep. HPLC requires a high eluent flow rate and the inner diameter of the column used in the prep. HPLC is usually larger. Required flow rate for prep. HPLC is usually between 10 to 100 ml/min (larger columns may require a flow rate of 500 ml to 1 liter/min). To prepare. HPLC, analytical pumps are designed for high pumping speeds and large volumes. The main modification that is required to work at a flow rate of 10-100ml/min is a larger piston head with a larger volume of liquid filled chamber.

3. Preparative Injector:

The Rheodyne injector should be capable of injecting a sample in the range of 0.1 to 100 mL volume under high pressure (up to 4000 psi). Also referred to as a rotary sampling valve. It involves four steps of operation. The preparative rheodyne injector and the rheodyne injector positions are shown in the figure

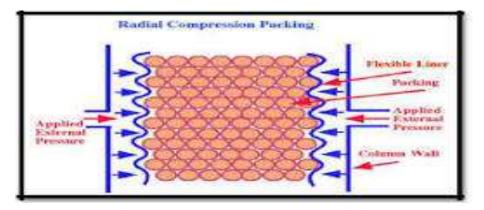


4. Preparative Columns: The role of the column is very important in the development of a reproducible preparation. HPLC method. Preparative columns should be able to withstand the high inlet pressure necessary to achieve the desired flow rate. The particle size of the packing material used for the column defines the performance of the prep. column. Because the preparative columns are wide, a sample distribution plate is placed to distribute the sample across the column. The distributor plate consists of a disk with radial slots.

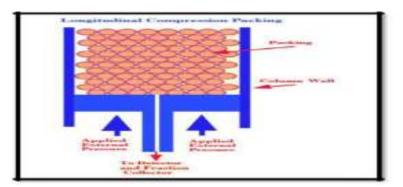
5. Packing of preparative column:

To pack the prep. HPLC columns efficiently, the pressure upto 650 bar or 9500 psi is necessary.

- 1. Dry Packing
- 2. Slurry Packing
- 1. Dry Packing: Dry packing is used when the particle size of the filling material is greater than 20 µm. It can be charged with knocking or sonic vibrations.
- 2. Slurry Packing: Liquid chromatography columns are packed with slurry when the particle size of the packing is less than 20 µm. The slurry is prepared by adding silica (2.2 g) in suitable solvents such as dibromomethane, tetrachloroethylene, carbon chloride, diiodomethane, etc.
- 3. Column packing techniques:
- 1. Radial compression packing:



2. Longitudinal Compression Packing:



6. Detector: Prep. chromatographic detectors can have very limited specifications. It may not be sensitive because the sample size and the solute concentration in the eluent are very large. Analytical detectors can be used for preparative purposes. In prep. The HPLC sample should be diluted with a large volume of mobile phase and then passed through the detector.

7. Fraction collector:

In prep. The HPLC diverter valve is used to divert the sample flow either to the waste or to the desired portion of the sample injected into the fraction container using the fraction collection needle.

Different fraction collection methods:

- 1. Mass collection of fractions
- 2. Collection of fractions based on vertex
- 3. Collecting factions based on time
- 4. Manual collection of fractions

APPLICATIONS OF PREPARATIVE HPLC:

- > Preparative HPLC is used in the pharmaceutical industry, to study drug stability and for quality control analysis.
- > Preparative HPLC is used in the purification of compounds in natural product chemistry.
- Micropurification and purity analysis of several different classes of compounds can be performed using preparative HPLC and pure compounds can be further used for advanced studies.
- > Preparative HPLC is used in the purification of by-products arising from the analysis of impurities.
- > Preparative HPLC is used in clinical testing for antibodies, urea and bilirubin.
- > Preparative HPLC is used to determine contamination and water pollution in environmental analysis.
- > Preparative HPLC is used in forensic analysis to quantify drugs and poisons in body fluids.
- Reported studies are as follows:

Sr	REPORTED STUDIES	CHROMATOGRAPHIC PARAMETERS	RESULTS
No.			
1	Isolation and purification of	1. Mobile phase: Methanol 20% - Trifluroacetic	About 110.7mg of heroin HCl was
	heroin from heroin street	acid in water 0.05%	obtained from 180mg of heroin
	sample	2. Column: C18 (250mm x 20mm, I.D. 15µm)	street sample which contains
		3. Flow rate: 20 ml/min.	156.15mg of heroin HCl with the
		4. Detector: SPD-M20A Photodiode- array	purity of about 99.52%
		detector	
2	Resolution of Gossypol:	1. Mobile phase: MeCN: 0.01 M phosphate	Multi-gram of Gossypol
	Analytical and Large-Scale	buffer (82: 18 v/v)	enantiomers are formed with high
	Preparative HPLC on Non-	2. Column: C18(33 cm x 22 mm I.D.)	purity
	Chiral Phases	3. Flow rate: 8.5 ml/min	
		4. Detector: Variable wavelength detector	

3	Identification, isolation and characterization of potential degradation product in lansoprazole drug substance	 Mobile phase: Water and acetonitrile 80:20 (v/v) Column: C18 (250mm long×9.4mm I.D.) Flow rate: 10 ml/min. Detector: Photodiode-array detector 	The isolation of impurity which is obtained as red color solid and the chromatographic purity was 96.0% by area percentage
4	Preparative high- performance liquid chromatographic purification and structural determination of 1-O-α-dglucopyranosyl-d- fructose (trehalulose)	 Mobile phase: Water Column: C18 columns (30 x 5.8 cm I.D.) Flow rate: 0.1 ml/min. Detector: Refractive-index detector 	With 98% purity, gram quantities of trehalulose can be conveniently obtained using preparative HPLC
5	Improved gram-quantity isolation of malto- oligosaccharides by preparative HPLC	 Mobile phase: Acetonitrilc- H2O (11:9) Column: Aminopropyl silica gel column (21.4 and 41.4 mm I.D. X 250 mm) Flow rate: I3 mL/min Detector: Refractive index detector 	Isolation of 1 g of oligosaccharide per hour. The purity of isolated malto-oligosaccharides was found to be at least 98%

Conclusion:

Preparative liquid chromatography is widely accepted as an excellent purification technique. Prep-HPLC with the same parts as analytical HPLC has some advantageous features compared to "low pressure" column chromatography systems, including particle size (prep-HPLC smaller, 3–10 mm) and large surface area results in a system with high resolution performance and can also be connected to some detectors such as diode array, refractive index, fluorescence, etc.

References:

- 1. Guo Z, Zheng H, et al, Forensic science international, 2012, (221): 120-124.
- 2. Matlin S, Belenguer A, et al, Journal of separation science, 1987, (10): 86-91.
- 3. Ramulu K, Rao B, et al, Journal of chemistry, 2013, (6): 274 283.
- 4. Cookson D, Cheetham P, et al, Journal of chromatography, 1987, (402): 265-272.
- 5. Hotchkiss A, Haines R, et al, Carbohydrate research, 1993, (242): 1-9.
- 6. Jadhav, Bharti Govind et al. "A Comprehensive Review for the Learners and Users: Preparative High Performance Liquid Chromatography." *International Journal of Chemical and Pharmaceutical Analysis* 1 (2014): 121-129.
- 7. Schulenberg-Schell, Helmut & Tei, Andreas. (2015). Principles in Preparative HPLC A Primer.
- M. Verzele, E. Geeraert, Preparative Liquid Chromatography, *Journal of Chromatographic Science*, Volume 18, Issue 10, October 1980, Pages 559–570, <u>https://doi.org/10.1093/chromsci/18.10.559</u>
- Kevin B. Hicks, Arland T. Hotchkiss, Chapter 15 Preparative HPLC of carbohydrates, Editor(s): Ziad El Rassi, Journal of Chromatography Library, Elsevier, Volume 66, 2002, Pages 505-534, ISSN 0301-4770, ISBN 9780444500618, https://doi.org/10.1016/S0301-4770(02)80040-9.
- S. Asha, P. Thirunavukkarasu, A. Mohamed Sadiq. Preparative HPLC Method for the Isolation of Compounds from Euphorbia hirta Linn. Ethanol Extract. Research J. Pharm. and Tech 2018; 11(6): 2541-2545. doi: 10.5958/0974-360X.2018.00469.9
- 11. https://www.chemistryworld.com/download?ac=142625
- 12. Tiebach, R.K.D., Schramm, M. Application of preparative HPLC in the chemical analysis of food ingredients. *Chromatographia* **13**, 403–407 (1980). <u>https://doi.org/10.1007/BF02261520</u>
- Wang, Xiaohong et al. "PREPARATIVE ISOLATION AND PURIFICATION OF CHEMICAL CONSTITUENTS OF BELAMCANDA BY MPLC, HSCCC AND PREP-HPLC." Journal of liquid chromatography & related technologies vol. 34,4 (2011): 241-257. doi:10.1080/10826076.2011.547058