

International Journal of Research Publication and Reviews

Journal homepage: www.ijrpr.com ISSN 2582-7421

Exploring Defect-Induced Anomalies in Oxide Nanoparticles' Magnetic Behavior

Kakarla Durga Rao¹, Dr. Veeresh Kumar²

¹Research Scholar , ²Research Guide Department of Physics, Sunrise University, Alwar, Rajasthan

ABSTRACT:

Oxide nanoparticles have garnered significant attention due to their unique properties and potential applications in various fields, including electronics, catalysis, and biomedical imaging. Among the intriguing properties exhibited by oxide nanoparticles, their magnetic behavior stands out as a topic of particular interest. In this paper, we delve into the phenomenon of defect-induced anomalies in the magnetic behavior of oxide nanoparticles. Defects, which can arise from lattice imperfections, doping, or surface modifications, play a crucial role in altering the magnetic properties of nanoparticles. This paper aims to provide an overview of the current understanding of defect-induced magnetic anomalies in oxide nanoparticles, including their underlying mechanisms, characterization techniques, and potential applications.

Introduction:

Oxide nanoparticles exhibit distinctive magnetic properties that can significantly differ from their bulk counterparts due to their small size, high surface-to-volume ratio, and quantum confinement effects. Defects within these nanoparticles can introduce local perturbations in the crystal lattice, leading to modifications in their magnetic behavior. This paper focuses on the exploration of such defect-induced anomalies in oxide nanoparticles' magnetic behavior.

Exploring defect-induced anomalies in oxide nanoparticles' magnetic behavior involves studying how the presence of defects in these nanoparticles can affect their magnetic properties. Oxide nanoparticles are materials with dimensions on the nanoscale that exhibit unique properties due to their small size and high surface-to-volume ratio. These properties can be further influenced by the introduction of defects, which are structural imperfections in the crystal lattice of the material.

In the context of magnetic behavior, defects can play a significant role in altering the magnetic properties of oxide nanoparticles. Some ways in which defects can influence magnetic behavior include:

Magnetic Ordering: Defects can introduce disruptions in the regular arrangement of atoms, leading to changes in the magnetic ordering of the material. This can result in modifications to the material's overall magnetic behavior, such as changes in the Curie temperature (the temperature at which a material's permanent magnetism changes to induced magnetism).

Spin Configurations: Defects can affect the alignment of electron spins in the material, which is crucial for its magnetic properties. Depending on the type and arrangement of defects, the spin configurations of electrons can be altered, leading to changes in the material's magnetic moment and susceptibility.

Magnetic Anisotropy: Defects can introduce strain or distortion in the crystal lattice, which can in turn influence the material's magnetic anisotropy. Magnetic anisotropy refers to the directional dependence of the material's magnetic properties. Defects can lead to changes in the preferred directions of magnetization, affecting the material's response to external magnetic fields.

Exchange Interactions: Defects can disrupt the exchange interactions between neighboring magnetic atoms. Exchange interactions are responsible for establishing the magnetic coupling between atoms and are crucial for determining the overall magnetic behavior of the material. Defect-induced changes in exchange interactions can lead to alterations in the material's magnetic properties.

Domain Structure: Defects can influence the formation and stability of magnetic domains within the material. Magnetic domains are regions in which the magnetic moments of atoms are aligned in a specific direction. Defects can affect the size, shape, and stability of these domains, impacting the overall magnetic behavior.

To explore these defect-induced anomalies, researchers typically employ a combination of experimental techniques and theoretical modeling. Techniques such as electron microscopy, X-ray diffraction, and various magnetic characterization methods are used to identify and characterize defects in oxide nanoparticles. Theoretical models, such as density functional theory (DFT) calculations, can provide insights into the electronic and magnetic properties of the materials and how they are affected by the presence of defects.

Diluted Magnetic Semiconductor Oxides (DMSO)

Despite the presence of intriguing methods for enhancing sample growth, the inclusion of magnetic Mn ions into III-V semiconductor crystals is constrained to approximately 0.1% when utilizing equilibrium growth parameters. Beyond this threshold, the likelihood of surface segregation and phase separation of dopant ions increases. In order to overcome the solubility issue, a non-equilibrium, low temperature molecular-beam-epitaxy (LT-MBE) technique was employed, resulting in the successful growth of (In,Mn)As and (Ga,Mn)As dilute magnetic semiconductor (DMS) ternary alloys with a Mn ion concentration exceeding 1%. Nevertheless, the initial publication in 1994 by Ohno et al. on the ferromagnetic transition in the p-type (In,Mn)As system noted a relatively low ferromagnetic transition temperature of Tc = 7.5 K45.

Since then, significant efforts have been made to increase the critical temperature in (III,Mn)V diluted magnetic semiconductors (DMSs), which has undergone various stages of anticipation and skepticism. Early experiments on the (In,Mn)As system revealed a significant correlation between the ferromagnetic transition and carrier localization. This relationship closely resembles the behavior observed in manganites, specifically the perovskite (La,A)MnO4 with A=Ca, Sr, or Ba. In the manganites, the well-established Zener double exchange process, accompanied by d-electron hopping between mixed valent Mn ions, is responsible for the observed ferromagnetism (Coey et al., 1999).

A comparable mechanism was also suggested to be accountable for facilitating the ferromagnetic interaction between Mn-Mn moments in the previously mentioned ferromagnetic (Ga,Mn)As samples, resulting in an increase in their critical temperatures (Tc) to approximately 50 K. This proposition was further supported by a groundbreaking theoretical ab-initio investigation of the ferromagnetic properties of (In, Mn)As. The research group from Tohoku University (Ohno, 1998) initially documented a significant increase in the critical transition temperature (Tc) to 110 K in p-type (Ga,Mn)As.

They also noted that this value aligned with the kinetic-exchange mechanism proposed by Zener for ferromagnetic coupling. The induction of ferromagnetism in these systems can be attributed to the indirect coupling between Mn d-shell moments, which is mediated by spin-polarized free-hole itinerant carriers. This coupling is commonly referred to as the Ruderman-Kittel-Kasuya-Yosida (RKKY) interaction (Dietl et al., 1997). Regrettably, the current debatability surrounding the applicability of Zener's model is evident.

This mechanism was initially proposed for transition metal ferromagnets, wherein itinerant carriers play a crucial role, while the d-electrons of transition metals exhibit a high degree of localization. The utilization of a localized moments model in Mn(d5) exchange-coupled through itinerant spband carriers offers a satisfactory explanation for the phenomenon of ferromagnetism observed in different Mn-doped IV-VI and II-VI diluted magnetic semiconductors (DMSs). It is noteworthy to mention that, in contrast to Mn-doped IV-VI and II-VI diluted magnetic semiconductors (DMSs), the incorporation of Mn in (III,Mn)V systems involves the substitution of the trivalent cation (such as Ga in GaMnAs). This substitution results in Mn acting as both an acceptor and a generator of magnetic moments.

Theoretical calculations utilizing Zener's kinetic-exchange model have indicated the possibility of observing ferromagnetism at room temperature in (Ga,Mn)As, particularly when the Mn content exceeds 10%. Despite the increasing number of experimental studies and optimistic forecasts, achieving further improvements in the critical temperature (TC) beyond the reported value of 110 K in (Ga,Mn)As has proven to be challenging for several years. In recent times, progress in molecular beam epitaxy (MBE) growth and the development of post-growth refinement techniques have enabled the suppression of various extrinsic effects, resulting in an increase in the critical temperature (Tc) of (Ga,Mn)As up to 174 K47.

The observed trend of elevated Curie temperature (Tc) in conjunction with the fabrication of high-quality (Ga,Mn)As epilayers aligns with the theoretical framework of the proposed Zener kinetic-exchange model. The objective was to cultivate diverse DMS materials with an elevated concentration of Mn ions in order to surpass the transition temperature. Based on a limited number of experimental and theoretical studies, it is possible to categorize (III,Mn)Sb diluted magnetic semiconductors (DMSs) as being similar to (Ga,Mn)As and (In,Mn)As DMSs.

However, theoretical calculations utilizing the kinetic-exchange model suggest that the critical temperatures (Tc) in antimonite systems are expected to be lower when compared to their analogous arsenide counterparts. The probable cause can be attributed to the diminished p-d exchange and reduced magnetic susceptibility (resulting from a smaller effective mass) of itinerant holes in the antimonite with a larger unit-cell. This hypothesis was subsequently validated by experimental researchers. Furthermore, the observed increase in the critical temperature (Tc) by 45% in the (In,Mn)Sb system, resulting from the application of external hydrostatic pressure, aligns with the predictions of the kinetic-exchange model.

The exploration of (III,Mn)P and (III,Mn)N as potential high Tc ferromagnetic semiconductors suggests that moving in the opposite direction in the periodic table is a viable approach. According to Dietl et al. (4000), the kinetic-exchange model accurately predicts a transition temperature that is significantly higher than room temperature in materials with a relatively smaller lattice constant, specifically in the case of (Ga,Mn)N. The solubility limit of manganese (Mn) in (gallium, manganese) nitride ((Ga,Mn)N) is significantly greater than in arsenides. This suggests that the inclusion of Mn ions at higher concentrations is feasible under or near equilibrium growth conditions.

Nevertheless, a comprehensive understanding of the magnetic interactions in Mn-doped phosphides and nitrides is still lacking, both in terms of

theoretical explanations and experimental investigations. The occurrence of significant charge fluctuations in the d-states can be attributed to the shift in the valence band edge towards the Mn d-level, as well as the increase in p-d hybridization resulting from the widening of the semiconductor band gap and the reduction in lattice constant.

Furthermore, as the ionicity of the host crystal continues to increase, there may be a transition from a d5 divalent acceptor to a d4 trivalent neutral impurity for dopant Mn ions. In both scenarios, it is imperative to subject the kinetic-exchange model put forth by Zener, which aims to elucidate the phenomenon observed in these materials, to meticulous examination.

Despite experimental reports of achieving transition temperatures close to 1000 K in (Ga,Mn)N samples, the exact cause of the high-temperature ferromagnetic phase remains uncertain. It is unclear whether this phenomenon can be attributed to the (Ga,Mn)N ternary alloy itself or to the presence of ferromagnetic nanoclusters or phase-separated metal precipitates embedded within the GaN lattice. Furthermore, there have been several reports discussing the synthesis of (Ga,Mn)N epilayers in both cubic and hexagonal crystal structures.

These studies have also observed the presence of ferromagnetism in both p-type and n-type (Ga,Mn)N. Additionally, the existence of multiple ordered phases within a single material further contributes to the intricate phenomenology observed in these wide bandgap diluted magnetic semiconductors (DMSs). The existing literature on the investigation of ferromagnetism in the (Ga,Mn)P system, specifically focusing on samples prepared through post MBE ion-implantation of Mn and subsequent annealing via rapid thermal or pulse laser melting, displays a significant amount of ambiguity in experimental reports and interpretation.

Types of Defects and Their Impact:

Various types of defects, such as vacancies, interstitials, and dopants, can exist within oxide nanoparticles. These defects can create localized magnetic moments, alter the crystal structure, and influence the magnetic interactions between nanoparticles. The paper discusses how different types of defects affect the overall magnetic behavior, including the emergence of ferromagnetic, antiferromagnetic, or even novel magnetic states.

Mechanisms of Defect-Induced Magnetic Anomalies:

The mechanisms behind defect-induced magnetic anomalies involve interactions between defect-induced magnetic moments and the surrounding lattice, as well as coupling between neighboring nanoparticles. The paper explores exchange interactions, spin-orbit coupling, and the role of defect-induced disorder in promoting magnetic frustration and novel magnetic phases.

Characterization Techniques:

To understand and control defect-induced magnetic anomalies, advanced characterization techniques are required. The paper covers techniques such as transmission electron microscopy (TEM), X-ray diffraction (XRD), electron paramagnetic resonance (EPR), and superconducting quantum interference device (SQUID) magnetometry. These techniques offer insights into the structural, electronic, and magnetic properties of oxide nanoparticles.

Applications and Future Prospects:

The ability to engineer and manipulate defect-induced magnetic anomalies opens up new avenues for applications. These include enhanced catalytic activity, improved magnetic data storage, and potential uses in medical imaging and targeted drug delivery. The paper discusses the potential impact of defect-engineered oxide nanoparticles in various fields and highlights emerging research directions.

Conclusion:

Defect-induced anomalies in oxide nanoparticles' magnetic behavior represent a promising area of research with profound implications across multiple disciplines. By elucidating the underlying mechanisms, characterizing these anomalies, and exploring their applications, researchers can harness the potential of defect engineering to tailor oxide nanoparticles' magnetic properties for specific needs.

REFERENCES:

1. M. Kavitha and C.Gopinathan and P.Pandi (4014). Synthesis and Characterization of TiO4 Nanopowders in Hydrothermal and Sol-Gel Method. Internat. Jour. of Advance. in Research & Tech., 4, 104-108.

2. M. Jlassi, M. Kandyla (4015). Journal of Alloys and Compounds, 646, 87–94.

3. M. Muccini (4006), Nature Materials, 5, 605.

4. R. Vijayalakshmi and V. Rajendran (4014).Synthesis and characterization of nano-TiO4 via different methods. Arch. of Applied Sci. Research,4,1184-1190.

5. R. Riasat and G. Nie (4016), Journal of Nanomat., 1-7.

6. S. R. Khan, S.Jamil, M.R Saeed, A. Janjua and R.A khera (4017). Synthesis of ferric Oxyhydrides nanoparticles and ferric oxides nanorods by reflux assisted co-orecipitation method and comparative study of their thermal properties, 4, 11.

7. S Basvaraja, D.S balaji, M. D bedra, D Raghunandan, P. M prithviraj swany and A Venkataraman (4011). Solvothermal synthesis and characterization of acicular α -Fe4O4 nanoparticles, Bull. Mater. Sci. 44, 1414–1417.

8. S. Valencia, J. M. Marín and G. Restrepo (4010), The Open Materials Science Journal, 4, 9-14.

9. T. T. Le Dang, M. Tonezzer and V. H Nguyen (4015). Hydrothermal Growth and Hydrogen Selective Sensing of Nickel Oxide Nanowires. Journal of Nanomaterials, 4015, 1-8.

10. U. Ashraf and B. Khan (4015).Synthesis and Characterization of NiO Nanopowder by Sol-Gel Method. Internat. Journal of Sci. and Research,4, 4405-4408.

11. W. Qin, C. Y. Ran Yi and U. Gao(4010). Hydrothermal Synthesis and Characterization of Single-Crystalline α-Fe4O4 Nanocubes. Journal of Nanomaterials,4011, 1-6.

12. X. Sun, H. Liu, J. Dong, J. Wei, Y. Zhang (4010), Catal. Lett., 145, 419.

13. Y. Li and J. Shi (4014), Advanced Materials, 46,4176–4405.

14. Z. Wang, Z.Wang, H. Wu, and X. W. Lou (4014), Scientific Reports, 4, 1491.

15. Aghazadeh, M Karimzadeh, I Maragheh, MG & Ganjali, MR 4018, 'Enhancing the supercapacitive and superparamagnetic performances of iron oxide nanoparticles through yttrium cations electro-chemical doping', Materials Research, vol. 41, no. 5.

16. Al Kiey, SA Ramadan, R & El-Masry, MM 4044, 'Synthesis and characterization of mixed ternary transition metal ferrite nanoparticles comprising cobalt, copper and binary cobalt–copper for high- performance supercapacitor applications', Applied Physics A: Materials Science and Processing, vol. 148, no. 6, pp. 1–14.

17. Alva, S Utami, RS Shyuan, LK Puspasari, I & Mohammad, AB 4016, 'Synthesis and Characterization of Toluene Sulfonic Acid (Tsa)-Doped Polypyrrole Nanoparticles: Effects of Dopant Concentrations', International Journal of Innovation in Mechanical Engineering and Advanced Materials, vol. 4, no. 1, pp. 1.

18. An, H Wang, Y Wang, X Zheng, L Wang, X Yi, L Bai, L & Zhang, X 4010, 'Polypyrrole/carbon aerogel composite materials for supercapacitor', Journal of Power Sources, vol. 195, no. 19, pp. 6964–6969.

19. Ansari, R 4006, 'Polypyrrole Conducting Electroactive Polymers: Synthesis and Stability Studies', E-Journal of Chemistry, vol. 4, no. 4, pp. 186–401.

20. Azharudeen, AM Karthiga, R Rajarajan, M & Suganthi, A 4040, 'Fabrication, characterization of polyaniline intercalated NiO nanocomposites and application in the development of non-enzymatic glucose biosensor', Arabian Journal of Chemistry, vol. 14, no. 4, pp. 4054–4064.

21. Azioune, A Slimane, A Ben Hamou, LA Pleuvy, A Chehimi, MM Perruchot, C & Armes, SP 4004, 'Synthesis and Characterization of Active Ester-Functionalized Polypyrrole-Silica Nanoparticles: Application to the Covalent Attachment of Proteins', Langmuir, vol. 40, no. 8, pp. 4450–4456.

22. Bachhav, SG & Patil, DR 4015, 'Study of Polypyrrole-Coated MWCNT Nanocomposites for Ammonia Sensing at Room Temperature', Journal of Materials Science and Chemical Engineering, vol. 04, no. 10, pp. 40–44.

23. Bhadrapriya, BC Varghese, AR Amarendra, G & Hussain, S 4018, 'Sb:SnO4 thin films-synthesis and characterization', AIP Conference Proceedings, vol. 1951, .

24. Bindu, P & Thomas, S 4014, 'Estimation of lattice strain in ZnO nanoparticles: X-ray peak profile analysis', Journal of Theoretical and Applied Physics, vol. 8, no. 4, pp. 144–144.

25. Bose, CSC & Rajeshwar, K 1994, 'Efficient electrocatalyst assemblies for proton and oxygen reduction: the electrosynthesis and characterization of polypyrrole films containing nanodispersed platinum particles', Journal of Electroanalytical Chemistry, vol. 444, no. 1–4, pp. 445–456.

26. Bose, P Ghosh, S Basak, S & Naskar, MK 4016, 'A facile synthesis of mesoporous NiO nanosheets and their application in CO oxidation', Journal of Asian Ceramic Societies, vol. 4, no. 1, pp. 1–5.