



Synthesis of Novel Thiazolidinone Derivative as a Potential Anti-Diabetic Activity

Deepali Gopal Pawar¹, Pramod. N¹

¹Department of Pharmaceutical Chemistry, A.G.M College of Pharmacy Varur, Hubli, Karnataka, India.

Email id: pramodnayanapalli@gmail.com

ABSTRACT

Thiazolidinone is a biologically important five-membered heterocyclic ring with almost all biological activities. Recent advances in the biological activities reported for 4-thiazolidinone derivatives, such as peroxisome proliferator-activated receptors γ binders, and antioxidants, anti-tuberculosis have been covered in this review. It includes the method of synthesis of thiazolidinone. It is synthesized by grindstone technology as a facile method. It mainly involves the green chemistry synthesis approach. Novel Thiazolidinone derivative is synthesized by simple synthesis of imines through nucleophilic acyl substitution reaction which involves a reaction between aniline and benzaldehyde in the presence of catalyst natural acid. The compound formed is treated with thioglycolic acid and 1,2-dioxane stirring the reaction at temperature 60°C for 4hrs to give thiazolidinones. The obtained thiazolidinones undergone conventional synthesis is refluxed for 12hrs to give the title compound. Thiazolidinone is screened for antidiabetic activity by *in vitro* glucose uptake assay using Pioglitazone as standard. The synthesized compound showed a very good potent activity towards anti-diabetic properties compared with standard pioglitazone drugs. Thiazolidinones are the class of oral agents for the treatment of type – 2 diabetes, improving insulin sensitivity and lowering blood glucose, free fatty acid, and triglycerides levels. The thiazolidinones are PPAR- γ (peroxisome proliferative-activated receptors) agonists.

KEYWORDS: Thiazolidinone derivatives, anti-diabetic activity,

1. INTRODUCTION

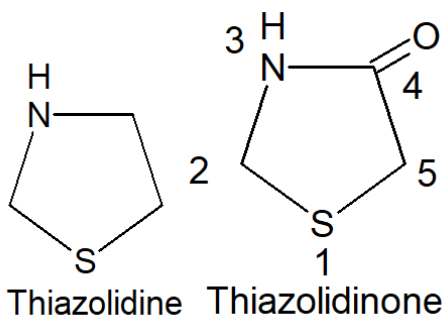
1.1. DIABETES:

Diabetes is a disorder where the blood glucose (sugar) level was higher than normal Blood glucose (sugar) level is known as **Hyperglycemia**.

Blood glucose levels are normally controlled by the hormone called insulin which is produced by beta cells of Langerhans in the pancreas (a gland which lies just below the stomach)¹

1.2. THIAZOLIDINONES-STRUCTURE

Thiazolidinones are derivatives of Thiazolidine with a carbonyl group at the 4-position.



1

2

Substituents in the 2-, 3-, and 5-positions may be varied, but the greatest difference in structure and properties is exerted by the group attached to the carbon atom in the 2-position (R and R' in 2 or X in 3).²

1.5. INTRODUCTION TO SYNTHESIS OF TITLE COMPOUND:

Grindstone technique: The titled compounds are prepared by the grinding technique which is one of the green synthesis approaches. Green chemistry is the utilization of a set of principles that reduces the use of hazardous chemical substances in the design, manufacture, and application of chemical products.⁶

Principle: The principle involved in the synthesis was a frictional force made between the mortar and pestle.⁷

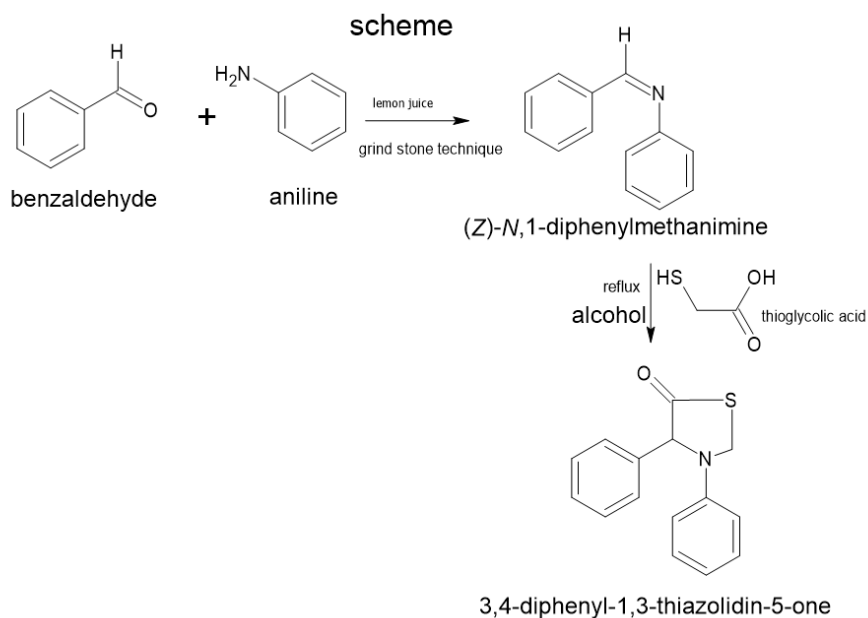
Application:

- Less hazardous chemical synthesis, with less energy and less time the compounds obtained.
- Solvent free synthesis.
- Cost effective
- Biocompatible.
- Biodegradable.
- Reliable and versatile.
- Convenient to handle.
- Simplicity.
- Safer and prevention of accidents.
- No use of higher temperature.
- Less wastage of chemicals (renewable).⁸
- Peroxisome proliferator-activated receptor γ binder

2. EXPERIMENTAL WORK

2.1. SYNTHETIC WORK

SCHEME OF WORK



General method for the Synthesis of Title Compound :

The procedure for the synthesis of compounds consists of two steps .

Step 1 : Synthesis of schiff base:

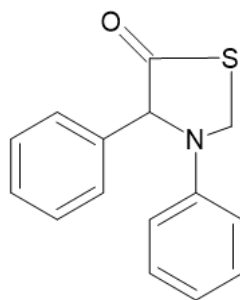
0.01 mol aniline and 0.01 benzaldehyde is taken in a mortar. The reaction mixture was undergone continuous stirring at room temperature by using lemon juice as a natural catalyst for 2 hrs. While trituration water can be added. Filtered and dried. To form the imines as a primary intermediate for the synthesis of thiazolidinones. This reaction involves the synthesis of Schiff base.

Step 2: Synthesis of thiazolidinone :

0.01 mol of substituted Schiff base and 0.01 mol of Thioglycolic acid and 1,2-dioxane was added and the reaction mixture was heated under reflux for 12 h. Concentrated, cooled and poured into the crushed ice. The solid thus separated was filtered, washed with water and recrystallized from ethanol. After the completion of reaction the reaction mixture was poured into crushed ice and the solid obtained was filtered, dried and recrystallized from ethanol.

The compound that was synthesized is represented as below :

IUPAC name: 3,4 -Diphenyl-1,3-Thiazolidin-5-One

**PHYSICAL CHARACTERIZATION:**

Solubility: Solubility of the synthesized title compound was tested using different solvents.

Melting point : The synthesized compound has the melting point at 84 °C.

2.2 ACTIVITY SCREENING**2.2.1. In vitro Antidiabetic activity:**

Name of analysis method: glucose uptake assay

Procedure:

3T3-L1 adipocytes, were seeded at a density of ~1500 cells per well in a 96-well plate, differentiated and maintained for another 10 days prior to use. To assay glucose uptake, adipocytes were starved in 100 μ l serum free adipocyte medium overnight (to enhance glucose uptake) then washed with PBS, followed by an incubation (40 min) in a glucose free medium (100 μ l Krebs-Ringer-Phosphate-HEPES (KRPH) buffer with 2 % BSA) then stimulated either with insulin (PGZ) (10 μ M), compounds (10 μ g/ml) or PBS. 10 μ l of 10mM 2-Deoxy glucose (DG) was added and the cells incubated for 20 min. The amount of glucose uptake was determined as per manufacturer's protocol using the Glucose uptake kit from Biovision (glucose uptake colorimetric assay kit, the 2-DG6P is oxidized to generate NADPH, which can be determined by an enzymatic recycling amplification reaction, color generated can be quantified colorimetrically at 412 nm.). The calculation was carried out keeping 100% glucose uptake for Pioglitazone (PGZ) was used as a standard drug.⁹

$$2\text{-DG uptake} = Sa/Sv \text{ (pmol/}\mu\text{l or nmol/ml or } \mu\text{M)}$$

Where: Sa is the amount of 2-DG6P (in pmol) in sample well calculated from Standard Curve.

Sv is sample volume (in 20 μ l) added into the sample well.

3. EXPERIMENTAL RESULTS**3.1 BIOLOGICAL ACTIVITY:****3.1.1. In-vitro Antidiabetic Activity:**

Antidiabetic activity of the synthesized derivatives was performed by the Glucose uptake assay and the results were tabulated below.

In-vitro Antidiabetic Activity:

Effect of compound PD-1) on 2-DGuptakein 3T3-L1 presence and absence of insulin:**Table no. 1**

S. No	Compound	OD(412)	2DG6P(pmol)	2-DGuptak(Pmol/μl)
1	Insulin(1microMol)	3.04	174	11.5
2	PD1	2.02	180	10.2
3	PD1+Insulin	10.7	460	23.4
10	Pioglitazone(10Micro.Mol)+Insulin(1Micro.Mol)	10.4	450	22.5
11	Pioglitazone(10uuMicro.Mol)	3.1	210	10.5

The results that are tabulated in Table clearly show that the synthesized thiazolidine shows the equipotent activity to that of insulin and pioglitazone as a standard. It also shows synergistic action when it is combined with insulin and pioglitazone. There it has shown very good antidiabetic activity. By removing the molecules of pioglitazone which is the standard drug for diabetes mellitus, a novel compound thiazolidinone can be synthesized and even that shows the equipotent activity to that of pioglitazone.

4. DISCUSSION OF RESULTS

In this present research work, novel derivatives of Thiazolidinone containing pyridine derivatives were synthesized in a two-step facile procedure with good yields. The compound purification was done by the recrystallization process. Thiazolidinone derivatives were screened for *In-vitro* Anti-diabetic activity based on their *In-silico* anti-diabetic activity.

***In-vitro* Anti-diabetic activity:**

The anti-diabetic activity of the title compounds was done by the Glucose uptake assay, the target is Peroxisome Proliferator-Activated Receptor Gamma (PPAR γ) which was performed by using Pioglitazone as the reference standard. Synthesized thiazolidinone shows the equipotent activity to that of insulin and pioglitazone as a standard. It also shows the synergistic action when it is combined with insulin and pioglitazone.

PPAR γ agonists that selectively modulate the receptor activity maintaining glucose homeostasis without the adverse effects associated with Thiazolidinone containing derivatives is a promising approach in diabetes research. In this context, the observed insulin-sensitizing effect through 2-DG uptake of compounds could be possibly attributable to mechanisms beyond PPAR γ activation.

5. CONCLUSION

Novel thiazolidinone derivative are synthesized, characterized, and screened for *in-vitro* antidiabetic using respective standards.

The results of *in-vitro* antidiabetic screening revealed that the synthesized compound showed good activity when compared to standard. This may be due to the fact that the target.

The results of *in-vitro* antidiabetic screening revealed that the synthesized derivative have good antidiabetic activity when compared to that of currently marketed drugs pioglitazone and insulin.

Further research need to be carried out to know relationship between structure of thiazolidinone derivative and biological activity.

6. ACKNOWLEDGEMENT:

The authors are thankful to **Rajiv Gandhi University of Health Sciences**, Bengaluru for providing the funding to carry out this work.

7. REFERENCES

- Graham L. Patrick. An introduction to medicinal chemistry.1,94.
- <http://www.diabetesinfo.org.nz/blog/wp-content/uploads/2015/10/PDF-Introduction-todiabetes.pdf>.
- Andres,C.J.;Bronson,J.J.;D'Andrea,S.V.Deshpande,M.S.;Falk,P.J.;Grantyoung.Bioorg.Med.Chem.Lett.2000.10.715.
- Jerry R. et al., Experimental And Clinical Pharmacology, Australian Prescriber, 2004, 27 (3) ,67-70.
- M.G.Vigorita, A.Chimirri,S.Grasso,G.Fenech,J.Heterocycle.chem.1979, (16),257-261.
- Evans, R.M., Barish, G.D., Wang, Y.X.PPARs and the complex journey to Obesity. Nature Mediccine 10, 355-361(2004).

-
7. Y.R. Sharma. Elementary Organic Spectroscopy.1980
 8. www.who.int/diabetes/global-report
 9. Amin et al., synthesis and preliminary evaluation of some substituted coumarins as anticonvulsant agents. Bio org.Med.Chem.2008,16,5377-5388.