

International Journal of Research Publication and Reviews

Journal homepage: www.ijrpr.com ISSN 2582-7421

Immobilization of protease on modified glass beads by salinization: its use in commercial activity

Hiral D. Trivedi¹, Dr. Dinesh S. Patel², Bijal D. Patel³

Research Student¹, Professor², Assist. Professor³

Shree P. M Patel College of P.G. Std. & Res. in Sci. Anand. Recognized Research Center for Ph.D. (Applied Chemistry, Forensic Science & M.Phil. in Chemistry)

DOI: https://doi.org/10.55248/gengpi.4.523.38787

ABSTRACT:-

In the present work immobilization of Protease enzyme on glass beads by covalent coupling was done. Glass beads were bought from scientific stationary were used. Protease was immobilized by covalent coupling. After immobilization process for further study, various parameters were optimized and carried out the comparison studies by using protease loaded glass beads. Furthermore, FTIR and SEM analysis were done to prove that the enzyme was immobilized on the support bead. In application protease of it was applied to purified the fruit juice i.e. Pineapple juice, watermelon juice and sugarcane juice. In addition, this study introduces a new achievement of the eco-friendly effect of immobilized enzyme, due to reuse full function of immobilized enzyme.

Keywords: Protease, Glass beads, Enzyme Immobilization, FTIR & SEM study of alcohol protease.

Introduction:-

Enzyme have high specificity and gives catalytic activity under mild conditions. Therefore, they have larger space of applications in commercial field. They have efficient and eco-friendly nature. Industrial applications of free enzymes are till poor because, their separation is difficult and reuse as well as low stability upon exposure to pH values, organic solvents and heat. On other hand, Immobilized enzymes provides many advantages as compared to free enzymes. They are reusability, stability and easy separation from the reaction solutions.

Now a days, many techniques are developed like physical adsorption, ion exchange, matrix entrapment, covalent bonding, micro capsulation have been largely applied in enzyme immobilization process on the bases of property of various support materials, like synthesized and also which are found from nature. Synthesized support like, hydrogels, HEMA, carrageenan, glass beads, agar, poly vinyl alcohol (PVA). Most important criteria for the application of biocatalyst like immobilized enzymes, different kind of immobilized microbial cells, and its operational stability.

In present research work investigation of different study of immobilized protease on glass beads, also worked on the stability and activity. Mainly focused on reusability and activity of protease as its advantages in commercial level. Beneficial results were achieved by this research work as used more reaction cycles rather than one reaction cycle.

2. Experimental

Materials

Glass beads which are normally used in chemistry lab, NaOH from organic lab, H_2O_2 from medical store, Concentrate H_2SO_4 which is used in chemistry lab, Protease (200U/mg) & 3-aminopropyl-trirthoxysilane(3-APTES) from sisco research laboratories Pvt. Ltd.

Characterization

FTIR Characterization was recorded on FT-IR Spectrometer. At CHARUSET UNIVERSITY Changa, Gujarat. SEM electron microscopic analysis was one by Scanning electron microscope at CHARUSET UNIVERSITY Changa.

Surface cleaning and activation process of glass beads:

For cleaning and surface activation the of surface of glass beads were added in solution of 30% V/V NaOH and 6% V/V H_2O_2 . After, then heating at 60°- 80° C for 15 minutes. On completing this step beads were filtered out from the solution, washed with distilled water, and dried. In next step beads were added in to the solution of 30% V/V H_2SO_4 . 6% V/V again heated for 60 minutes at 90° C. This is the pre-conditioning method provides suitable surface with enough amount of silanol group for silan derivation to form siloxane covalent bonds. Formation of Si-O-Si bonds.

21 gm of glass beads from activated glass beads were taken, 150 ml of dry toluene and 5 ml of 3 -aminopropyl-triethoxysilan were added into reaction under nitrogen atmosphere. Then the was stirred under reflux for 15 h. afterwards, amino functionalized glass beads were separate out from the solution and kept in to the oven at 110° C for 2 h. for achieved complete

Immobilization of protease on glass beads:

After done all the cleaning and surface activation process glass beads were slowly stirred in reaction flask which contains 20% V/V glutaraldehyde solution. Stirring process was done for 2 h. at room temperature approx. 23 - 25° C. The glass beads were than immersed in protease solution which was diluted with sodium acetate buffer ($_{p}$ H=4) with ratio of 1:4. (6 ml enzyme solution in 24 ml of buffer solution) at 4° C. Then beads were used for further measurements and application.

Activity yield = Activity of immobilized protease ×100 Specific activity of free enzyme

Preparation of juice.

Freshly picked up pineapples were taken and kept at normal room temperature. Pineapple was washed properly and outer skin was removed from it, then fresh juice was made by normal mixture without adding any extra things. After complete filter process, 150 ml of juice was separate out in a beaker to perform assay method.

Treatment of immobilized Protease to check reusability

5 ml of juice from was taken in test vial; 0.5 ml of starch solution was added. After adding beads of immobilized enzyme contains 0.05 g of enzyme and phosphate buffer solution ($_{p}$ H=6.9) was incubated at room temperature for 10 minutes. 2 ml of 3, 5 -dinitrosalicylic acid was added and vail was incubated in boiling water bath for 10 minutes reducing sugar was determine by spectrophotometer at 420 nm.

pH activity profile

As enzymes consist of protein, the catalytic activity is markedly affected by environmental conditions, especially the pH of the aqueous medium. Thus, information on changes in pH-activity behavior caused by the immobilization of enzymes is useful for an understanding of the structure-function relationship of enzyme protein. Hence, the activity of the free and immobilized protease has been measured by incubating free and immobilized enzymes at 27 °C for 30 min in the 50 mm phosphate buffers of different pH ranging from 5 to 9 and using ethanol as a substrate. The absorbance of the reaction mixture was measured at 420 nm and correlated to the concentration of the enzyme. From the calibration, plot activity of the enzyme was determined.

Thermal stability

Because of the immobilization of enzyme the heat stability is enhanced, it is advantageous for the industrial application of immobilized enzymes and is thus important in determining the feasibility of immobilized enzymes for a particular application. Therefore, the thermal stability of free and immobilized enzymes was investigated. Free and immobilized enzymes were placed in the optimum pH buffer and incubated at different temperatures (30 to 70 °C) for different time intervals, activity of the enzyme was then determined as described earlier Thermal deactivation constant (Kd) was calculated by using the following equation :

Ln At = In Ao - Kd (t)

Where 'Ao' is the initial activity and 'At' is the activity after heat-treat for minutes.

3. Result and Discussion

pH Outline

Every enzyme has an optimum pH at which it shows optimum activity. **Figure 1**shows the pH activity of free and bounded. We have observed that free and entrapped enzyme was showing a maximum of 5 to 8 pH. During immobilization, results shows that there is no conformational change.

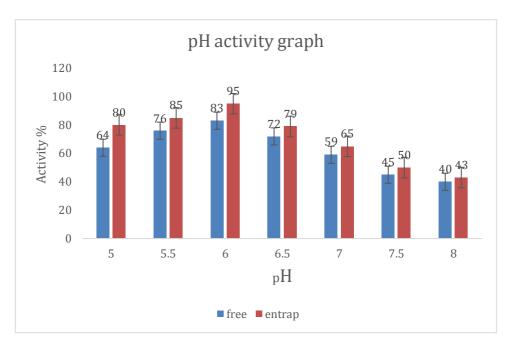


Fig-1 pH activity graph of immobilized protease on glass beads

Thermal Stability:

Enzymes is temperature-dependent. When the temperature increases enzyme reactivity increases and beyond a definite limit the enzyme, gets deactivated. **Figure 2**shows that entrapped enzyme show better thermal stability compared to free enzyme. The entrapped enzyme showed better thermal stability as they are encapsulated within the beads.

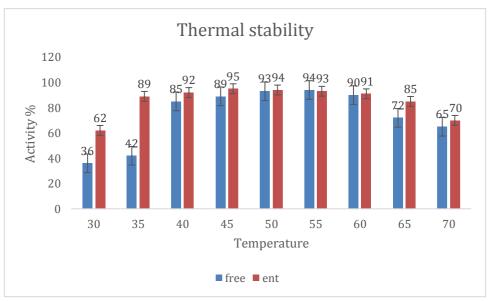


Fig-2 Thermal stability graph of immobilized protease on glass beads

FTIR analysis

FTIR analysis of bead was done. FTIR analysis shows the chemical presence of specific chemical groups of enzyme. Spectra were recorded in the spectral range of $4000 - 500 \text{ cm} \cdot 1.\alpha$ -amylase have amide group in its structure. Pease peak pears at 1651.58 m⁻¹ shows stretching of -C=O and 1026.07 cm-1 shows the bond starching vibration of amide. 3395.48 cm-1 shows –OH starching vibration.

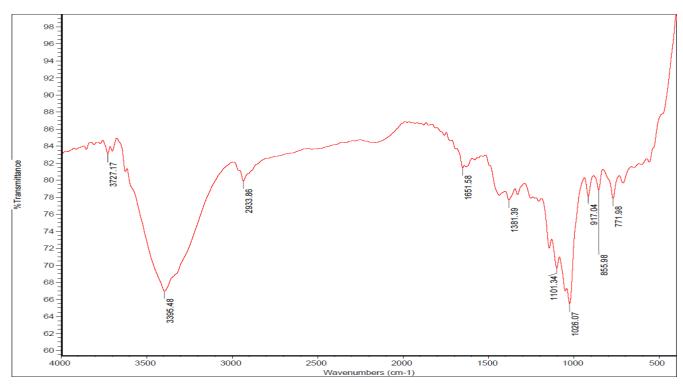


Fig-3 FT-IR of immobilized protease on glass bead.

SEM analysis (Scanning electron microscopy)

SEM analysis of bead in which have enzyme was entrapped, purpose of this study was to obtain a topographical Characterization of the support beads. SEM photographs were taken using a scanning electron microscope, at required magnification at room temperature. Distance of working 9.5 mm was maintained, and the acceleration voltage used was 20.00 KV

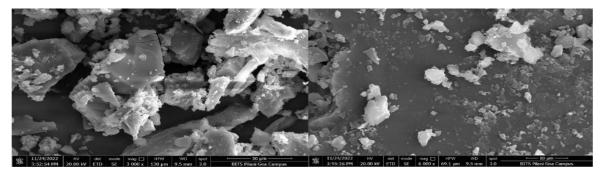


Fig-4 SEM image of immobilized protease on glass beads

Storage stability

For study of residual activities of the free and immobilized enzymes stored at room temperature (30 $^{\circ}$ C) were determined and the activities were expressed as percentage retention of their residual activities at different times. Immobilized beads were also kept at 5 $^{\circ}$ C for 45 days and residual activity was examine after every 7 days.

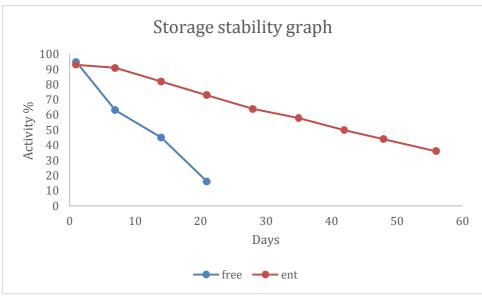


Fig-5 Thermal stability graph of immobilized protease on glass beads

Use of immobilized Protease in Juice purification.

Fresh pineapple juice, watermelon juice, sugarcane juice weaken. For this 5ml of juice was taken and added to the vial. Then entrapped enzymes beads were added in it after this in next step 5 ml of buffer solution was added in the test vial, and 1 ml of casein solution was added. Then vial was incubated in a shaker incubator at 35 °C for 30 min. after incubation 5 ml of carbonate and 1 ml of folin's reagent were added. And again kept the test vial for 30 min at 35 °C. Practical was performed with a free enzyme and with entrapped enzyme. OD (optical density) was carried out for further calculation at 420 nm.

Reusability of bead:

In industries and bio-chemical reaction, reusability of entrapped enzyme has great importance. It was checked by using beads in the assay method in place of free enzyme solution. The reusability of entrapped enzyme beads was checked and it was founded as shown in. **Figure-5** bounded enzyme employed 75% of its enzyme reactivity after 5 or 6 rotations and 50-40 % activity after 8 cycles showing the advantage of immobilized enzyme and which increases its applicability.

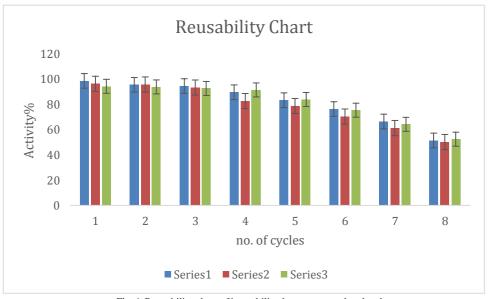


Fig-6 Reusability chart of immobilized protease on glass beads.

Acknowledgement:

I am grateful to my guide **Prof. (Dr.) Dinesh S Patel, Mrs. Bijal D Patel** research scholar and assistant professor and my institute Shree P. M Patel college of P.G. Std. & Res. in Sci. Anand. Recognized Research Center for Ph.D. (Applied Chemistry, Forensic Science & M.Phil. in Chemistry) for their help and cheering for my work. I would also like to thank Mr. Bipin Vakil the owner of our Medicare group for providing me with such a big opportunity in my life and for making the research facilities available in our institute.

REFERENCES:-

- 1.
 ^ Rawlings
 ND,
 Barrett
 AJ
 (February
 1993). "Evolutionary
 families
 of
 peptidases". The

 BiochemicalJournal. 290 (Pt1):205218. doi:10.1042/bj2900205. PMC 1132403. PMID 8439290.
 Finitian of the state of the
- ^ Jump up to:a b Rawlings ND, Barrett AJ, Bateman A (November 2011). "Asparagine peptide lyases: a seventh catalytic type of proteolytic enzymes". The Journal of Biological Chemistry. 286 (44): 38321–38328. doi:10.1074/jbc.M111.260026. PMC 3207474. PMID 21832066.
- 3. ^ Jump up to:a b Rawlings ND, Barrett AJ, Bateman A (January 2010). "MEROPS: the peptidase database". Nucleic Acids Research. 38 (Database issue): D227–D233. doi:10.1093/nar/gkp971. PMC 2808883. PMID 19892822.
- ^A Mitchell RS, Kumar V, Abbas AK, Fausto N (2007). Robbins Basic Pathology (8th ed.). Philadelphia: Saunders. p. 122. ISBN 978-1-4160-2973-1.
- ^ Rodriguez J, Gupta N, Smith RD, Pevzner PA (January 2008). "Does trypsin cut before proline?". Journal of Proteome Research. 7 (1): 300– 305. doi:10.1021/pr0705035. PMID 18067249.
- ⁶ Renicke C, Spadaccini R, Taxis C (2013-06-24). "A tobacco etch virus protease with increased substrate tolerance at the P1' position". PLOS ONE. 8 (6): e67915. Bibcode:2013PLoSO...867915R. doi:10.1371/journal.pone.0067915. PMC 3691164. PMID 23826349.
- ^ van der Hoorn RA (2008). "Plant proteases: from phenotypes to molecular mechanisms". Annual Review of Plant Biology. 59: 191– 223. doi:10.1146/annurev.arplant.59.032607.092835. hdl:11858/00-001M-0000-0012-37C7-9. PMID 18257708.
- ^Zelisko A, Jackowski G (October 2004). "Senescence-dependent degradation of Lhcb3 is mediated by a thylakoid membrane-bound protease". Journal of Plant Physiology. 161 (10): 1157–1170. doi:10.1016/j.jplph.2004.01.006. PMID 15535125.
- ⁶ Sims GK (2006). "Nitrogen Starvation Promotes Biodegradation of N-Heterocyclic Compounds in Soil". Soil Biology & Biochemistry. 38 (8): 2478–2480. doi:10.1016/j.soilbio.2006.01.006.
- 10. ^ Sims GK, Wander MM (2002). "Proteolytic activity under nitrogen or sulfur limitation". Appl. Soil Ecol. 568: 1-5.
- 11. ^ Tong L (December 2002). "Viral proteases". Chemical Reviews. 102 (12): 4609–4626. doi:10.1021/cr010184f. PMID 12475203.
- 12. [^]Skoreński M, Sieńczyk M (2013). "Viral proteases as targets for drug design". Current PharmaceuticalDesign. 19 (6):11261153. doi:10.2174/13816128130613. PMID 23016690.
- ^ Kurt Yilmaz N, Swanstrom R, Schiffer CA (July 2016). "Improving Viral Protease Inhibitors to Counter Drug Resistance". Trends in Microbiology. 24 (7): 547–557. doi:10.1016/j.tim.2016.03.010. PMC 4912444. PMID 27090931.
- ^ Jump up to:a b Giménez MI, Cerletti M, De Castro RE (2015). "Archaeal membrane-associated proteases: insights on Haloferax volcanii and other haloarchaea". Frontiers in Microbiology. 6: 39. doi:10.3389/fmicb.2015.00039. PMC 4343526. PMID 25774151.
- [^] Maupin-Furlow JA (December 2018). Robinson NP (ed.). "Proteolytic systems of archaea: slicing, dicing, and mincing in the extreme". Emerging Topics in Life Sciences. 2 (4): 561–580. doi:10.1042/ETLS20180025. PMC 7497159. PMID 32953999.
- [^]Barrett AJ, Rawlings ND, Woessnerd JF (2004). Handbook of proteolytic enzymes (2nd ed.). London, UK: Elsevier Academic Press. ISBN 978-0-12-079610-6.
- 17. ^ Hooper NM, ed. (2002). Proteases in biology and medicine. London: Portland Press. ISBN 978-1-85578-147-4.
- ^A Feijoo-Siota L, Villa TG (28 September 2010). "Native and Biotechnologically Engineered Plant Proteases with Industrial Applications". Food and Bioprocess Technology. 4 (6): 1066–1088. doi:10.1007/s11947-010-0431-4. S2CID 84748291.
- 19. ^ Southan C (July 2001). "A genomic perspective on human proteases as drug targets". Drug Discovery Today. 6 (13): 681–688. doi:10.1016/s1359-6446(01)01793-7. PMID 11427378.
- ^ Puente XS, López-Otín C (April 2004). "A genomic analysis of rat proteases and protease inhibitors". Genome Research. 14 (4): 609–622. doi:10.1101/gr.1946304. PMC 383305. PMID 15060002.