



## **A Review on: Antibacterial Activity of Ashwagandha**

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### **ABSTRACT:-**

The use of plants in the treatment of burns, ringworm and infecting diseases is common in traditional medicine. The development of new anti-microbial agents against opposing germs increases interest. Therefore, methanolic extracts from various parts of the four medicinal plants used locally in traditional medicine were tested for anti-bacterial task. It was found that numerous of the plants extracted from the plants studied had anti-bacterial and antifungal properties. Methanolic extracts of the plant *Azadiracta indica*, *Acacia nilotica* and *Witania somnifera* have shown significant anti-bacterial task against *Bacillus subtilis*, *Escherchia coli*, *stphaylococcus aureus* and *pseudomonas fluoescence*. *Azadiracta indica* and *A.tinolica* have shown significant antifungal task against *A. flavus*, *Ziziphus mauritiana*. The rhizome extract of *curcuma longa* showed significant task against all tested bacteria and showed high fungal task against *Fusarium verticillioides*.

Keywords: *Azadiracata*, *Somnifera*, Extract, Anti-bacterial, Germs, Rhizome

### **INTRODUCTION:**

*Somnifera* some heavy metals exert serious toxicities even at low intake levels.

However, heavy metals such as copper, iron, cobalt, and chromium in small concentrations are essential elements in the human body. Therefore, it is necessary to detect and remove heavy metals from our food and water resources. Conventional approaches for heavy metal detection from complex matrices include atomic absorption spectroscopy, neutron activation analysis, X ray fluorescence spectrometry, energy dispersive X-ray fluorescence, inductively coupled plasma mass spectrometry, inductively coupled plasma atomic/optical emission spectrometry, and flame atomic absorption spectrometry. Although they can generally achieve satisfactory detection sensitivity, they are inevitably challenged with heavy equipment, high cost, and lab-only and complicated operation. Classical methods for heavy metal removal from wastewater include precipitation, ion exchange, reverse osmosis, membrane filtration and oxidation. However, their effectiveness is limited due to disadvantages of sludge contamination, high pH sensitivity, and corrosiveness. Nanotechnology refers to employing multidisciplinary techniques integrating principles of physics, chemistry, engineering, and biology to build nanoscale materials or devices. With the upsurge of nanotechnology, nanomaterials have been delicately designed and fabricated for heavy metal detection and removal, exerting numerous advantages as compared to the previous methods.

### **Supramolecular chemistry:**

Supramolecular chemistry refers to the branch of chemistry concerning chemical systems composed of a discrete number of molecules. The strength of the forces responsible for spatial organization of the system range from weak intermolecular forces, electrostatic charge, or hydrogen bonding to strong covalent bonding, provided that the electronic coupling strength remains small relative to the energy parameters of the component. While traditional chemistry concentrates on the covalent bond, supramolecular chemistry examines the weaker and reversible non-covalent interactions between molecules.

These forces include hydrogen bonding, metal coordination, hydrophobic forces, van der Waals forces, pi-pi interactions and electrostatic effects. Important concepts advanced by supramolecular chemistry include molecular self-assembly, molecular folding, molecular recognition, host-guest chemistry, mechanically-interlocked molecular architectures, and dynamic covalent chemistry. The study of non-covalent interactions is crucial to understanding many biological processes that rely on these forces for structure and function. Biological systems are often the inspiration for supramolecular research. Supramolecular chemistry refers to the branch of chemistry concerning chemical systems composed of a discrete number of molecules. The strength of the forces responsible for spatial organization of the system.

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***Mechanism of supramolecular chemistry:***

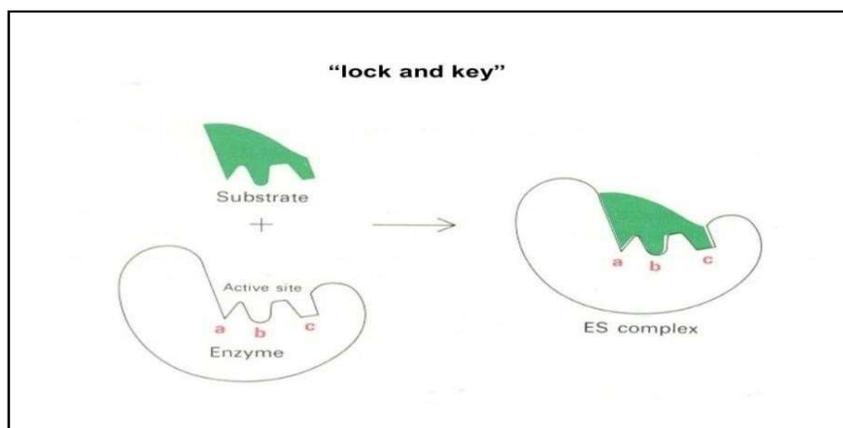


Figure No.01 mechanism of supramolecular Chemistry

The basis of supramolecular chemistry is the binding or complexation event, which is the action of the host molecule binding with a guest molecule to form a host-guest complex or supramolecule. The host is an enzyme or synthetic cyclecompound possessing a central hole or cavity. The guest is monoatomic cation or simple inorganic anion. In more complex biological interactions, the guest is called as substrate and can be a food particle, hormone, pheromone or neurotransmitter. Supramolecular chemistry is characterized by the specificity and selectivity of its reactions. The term used to describe this is molecular recognition, as if the reactions will only happen if the molecule recognizes each other. Therefore, a supramolecular interaction can only happen when the host and guest complement.

***Supramolecular nanomedicine***

Supramolecular chemistry enabling molecules and molecular complexes binding through non-covalent bonds allows nanomedicines to serve their desirable function to deliver drugs at the right time and the right place with minimal invasiveness. Supramolecular nanomedicine is the application of nanosupramolecules to the human health and disease and its main applications include diagnosis and therapy, drug and gene delivery, and tissue engineering.

Nanoparticles with different structures obtained by assembling supra amphiphiles are promising candidates for functional therapeutic platforms combining imaging and therapeutic capabilities. Encapsulation in supramolecular nanocarriers such as polymeric micelles, polymeric vesicles, layer-by-layer assembly, and porphyrins has the potential to deliver imaging and therapeutic drugs to the sites of action in the body. Hybrid supramolecular nanostructures of organic and inorganic molecules show promising potential in nanomedicine.

Research is progressing towards rapid development on supramolecular nanotheranostic devices. Moreover, supramolecular nanoparticles exhibit low-toxicity, low-immunogenicity, nonpathogenicity, and in vivo degradability.

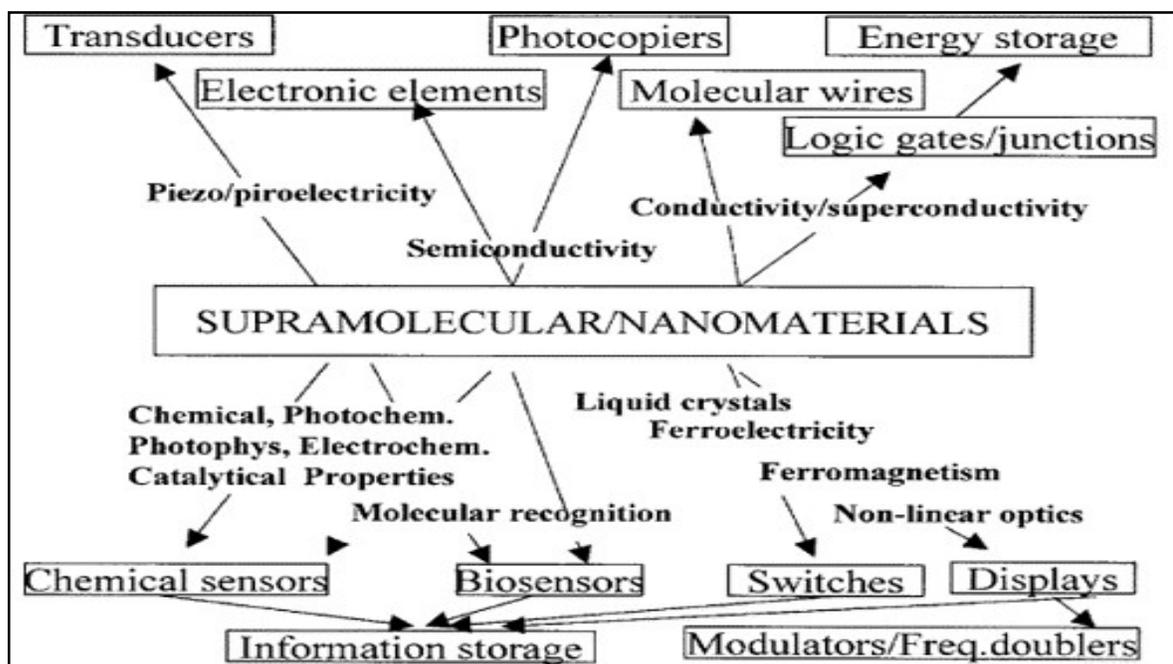


Figure no.02 Diagrammatic Representation of supramolecular nanomaterials.

#### Nano technology pre concentration separation Carbon-Based Nanomaterials

Carbon-based nanomaterials were initially applied in the electronics industry owing to their extraordinary and electrical properties. Nevertheless, some other exceptional properties they exhibited, such as a large surface area, ease of chemical or physical modification, ability of removing both organic, and inorganic pollutants have made carbon-based nanomaterials potential alternatives for treating waste water. Here, two major carbon-based nanomaterials are mainly presented—carbon nanotubes-based and graphene-based nanomaterials.

#### Carbon Nanotubes

Carbon nanotubes (CNTs) have been extensively investigated during the last decades and they were reported to exhibit many exceptional properties, including optical, electronic, vibrational, mechanical and thermal properties.

There have been numerous reports on their applications for the removal of heavy metals from waste water. Carbon nanotubes, basically divided into single-walled CNTs (SWCNTs) and multi-walled CNTs (MWCNTs).

#### Silica-Based Nanomaterials

Silica-based nanomaterials are another kind of important nanomaterial for removing heavy metals due to their properties, such as non-toxicity and excellent surface characteristics.

Nanosilica can be surface modified by groups like  $-NH_2$ ,  $-SH$ , etc, or serve as the support of nanocomposites.

**Zero-Valent Metal-Based Nanomaterials** Zero-valent metal nanoparticles have exhibited their potential in water treatment and remediation in recent years. For example Ag nanoparticles have been used to disinfect waste water due to their antimicrobial ability.

Nanosized zero-valent zinc was reported to have an excellent degradation ability towards dioxins. As for the heavy metal ions treatment, zero-valent iron was most extensively investigated and mainly discussed in this part.

#### Zero-Valent Iron nanomaterials

zerovalent iron (nZVI) is a composite consisting of Fe and ferric oxide coating. It has received increased attention as an adsorbent to treat various kinds of heavy metals, such as Hg, Cr, Cu, Ni, Cd, etc. since it came out. Basically, Fe provides the reducing ability while the ferric oxide shell offers

the sites of reactive and electrostatic interaction with heavy metals.

#### Ag Nanoparticles

Unlike nZVI, reports on other metallic nanoparticles which could be used to remove heavy metals are not sufficient.

There have been several reports about the interaction between Ag nanoparticles and Hg. Although the reactivity between Hg and the bulk silver is not high, Ag nanoparticles can exhibit a higher reactivity because their reduction potential of Ag decreased with the diminution of the particle size. A novel silver nanoparticle-based adsorbent was synthesized by coordinating Ag with mercaptosuccinic acid (MSA). Two different materials were prepared and investigated by varying the ratio of Ag to MSA. The results demonstrated that 1:6 Ag MSA had a higher removal capacity towards Hg compared with the common adsorbents. Furthermore, it stated that the cost for removing Hg by employing Ag@MSA was competitive, indicating that Ag@MSA could be a promising alternative in the removal of Hg.

#### Metal Oxide-Based Nanomaterials

Nanosized metal oxides possess many exceptional properties, such as a high removal capacity and selectivity towards heavy metals. Thus, they have great potential as promising adsorbents for heavy metals. Generally speaking, metal oxides-based nanomaterials include nanosized iron oxides, manganese oxides, zinc oxides, titanium oxides, aluminum oxides, magnesium oxides, cerium oxides, and zirconium oxides, etc.

#### Iron Oxides Based Nanomaterials

Iron oxide-based nanomaterials have received increasing attention in removing heavy metals from wastewater these years. Iron is the fourth most abundant element in the Earth's crust. The abundance of the iron element and the simplicity of synthesizing iron oxides contribute to the wide investigations on iron oxide-based nanomaterials. The most investigated iron oxides are  $\alpha$ -FeOOH, hematite ( $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>), maghemite ( $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>), magnetite (Fe<sub>3</sub>O<sub>4</sub>), and hydrous iron oxides (HFO).

#### Manganese Oxides-Based Nanomaterials

Manganese oxides nanoparticles have also been reported to remove heavy metals from wastewater and nano sized manganese dioxide as well as hydrous manganese oxide (HMO). Nanocrystalline manganese oxide was found to have a high surface area which contributes to its good adsorption performance. Also, M-O<sup>δ+</sup> and M-O<sup>δ-</sup> units on the surface of manganese oxide would help the absorption of metal ions. Wang et al. synthesized a novel dumb bell-like manganese dioxide/gelatin and investigated its adsorption performances towards Pb and Cd. The batch adsorption study showed that the maximum adsorption capacities towards Pb and Cd were 318.7 and 105.1 mg respectively by calculating from the Langmuir model.

#### Zinc Oxides-Based Nanomaterials

adsorbents for heavy metals due to their high surface area, low cost and extraordinary removal capacity. Nanosized zinc oxides have been reported to treat Cr, Cu, Ni and Al. studied the removal performances of Zn on particles towards Zn, Zinc oxide nanoparticles have gained their popularity as Cd and Hg by using batch method.

#### Titanium Oxides-Based Nanomaterials

Titanium oxides are extensively reported to photo degrade organic pollutants as effective photocatalytic. There are also some reports on their applications on the heavy metal removal. had prepared nano-titania with mesoporous structure via a rapid surfactant-free approach and investigated its adsorption effect towards dichromate with Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> equilibrium concentrations varying from 20–300 mg. The result indicated that the maximum adsorption capacity of the synthesized TiO<sub>2</sub> towards Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> was 26.1 mg·g<sup>-1</sup>, which was 12.6 mg·g<sup>-1</sup> for Cr. This nano-titanium adsorbent showed a higher uptake capability towards Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> than that of the previously reported adsorbent.

#### Aluminum Oxides-Based Nanomaterials

Aluminum oxides-based nanomaterials are another kind of widely used metal adsorbent towards heavy metals with the advantages of a low manufacturing cost and high decontamination efficiency. Aluminum oxide has several crystalline structures such as  $\alpha$ ,  $\gamma$ ,  $\theta$ ,  $\eta$  etc.

#### Magnesium Oxides-Based Nanomaterials

Magnesium oxide nanoparticles have many advantages as adsorbents for heavy metals, including an extraordinary adsorption capacity, low cost, nontoxicity, abundance, and environmentally friendly character. Moreover, MgO nanoparticles are also equipped with an excellent antibacterial ability towards Gram-positive and Gram-negative bacteria as well as bacterial spores. Research has indicated that Cd, Pb, and Escherichia coli could be removed from water simultaneously via MgO nanoparticles synthesized by the sol-gel method.

#### Cerium Oxides-Based Nanomaterials

Nanosized cerium oxide (CeO<sub>2</sub>), a non-harmful rare-earth oxide in which Ce is quadric valence, have been applied in many areas such as photo catalysis and sensing, UV blocking, water treatment, etc. The crystalline size, bulk density, porosity, and surface area of CeO<sub>2</sub> were reported to have great influence on their activity, stability, dispersion behavior, and selectivity, thus influencing the effect of heavy metals removal. investigated the adsorption effect of CeO<sub>2</sub> nanoparticles towards Cr from water. They synthesized CeO<sub>2</sub> nanoparticles had an average size of 12nm and surface BET area of 65 m<sup>2</sup>·g<sup>-1</sup>. The maximum adsorption capacity for Cr was 121.95 mg·g<sup>-1</sup> in this work with the initial Cr concentration of 80 mg/L, indicating a good choice for removing low amounts of Cr from water.

#### Zirconium Oxides-Based Nanomaterials

Nanosized zirconium oxides are another kind of promising metallic oxides adsorbent which can be used to remove heavy metals in wastewater. Their advantages are that they have plenty of -OH on their surfaces and possess large surface areas. Moreover, nanosized zirconium oxides own great chemical stabilities and exhibit excellent adsorption affinities towards heavy metals like Pb, Zn and Cd. Zirconium oxides nanomaterials are mainly comprised of nanosized zirconia and hydrous zirconia (HZO) based nanomaterials. investigated the removal effects of a series of transition-metals which had mesoporous structures towards Cr solutions, including TiO<sub>2</sub>, ZrO<sub>2</sub>, HfO<sub>2</sub>, Nb<sub>2</sub>O<sub>5</sub>, and Ta<sub>2</sub>O<sub>5</sub>. The synthesized metal oxides all exhibited large surface areas and were comprised of partially combined homogeneous nanoparticles with crystalline cores and amorphous shells.

#### Nanocomposite Nanomaterials

Despite the fact that each kind of nanomaterial has their own advantages, their respective drawbacks can not be neglected. For example, it is difficult for suspend uniformly in different solvents, while nZVI are prone to be oxidized.

#### Inorganic-Supported Nanocomposites

Inorganic supports of nanocomposites which are used for heavy metals removal are mainly consisted of activated carbon (AC), CNTs, and some natural materials such as bentonite, montmorillonite, zeolite, and so on. AC is one of the most effective, economic and simplest adsorbent for pollutants in the aqueous solutions. There have been some reports on the AC-supported nanocomposites which were used to remove heavy metals from water and these composites exhibited great potential for removing Cr, Pb, Cd etc. CNTs-supported nanocomposite is one kind of nanomaterial mainly supported on CNTs. Chitosan is one of the most widely used polymer to modify CNTs to prepare this kind of nanocomposites. prepared multi-walled CNTs/chitosan nanocomposite by sonicating the chitosan and CNTs suspension and then cross linking them with glutaraldehyde.

#### ***Organic Polymer-Supported Nanocomposites***

Polymeric hosts have many extraordinary properties, such as excellent mechanical strength, tunable functional groups, feasible regeneration, environmental soundness, and a degradable characteristic which make organic polymers a competitive option of hosts for nanocomposites. polymer-supported nanocomposites are comprised of two types, synthetic organic polymer-supported nanocomposites and biopolymer-supported nanocomposites

#### ***Magnetic Nano composites***

Magnetic nanocomposites are one type of peculiar nanomaterial which have been receiving increasing attention due to their easy separation ability. Magnetic nanocomposites are mostly based on magnetic iron and iron oxides. The fabrication of these magnetic nano composites could be achieved mainly through three approaches Surface modification of magnetic iron/iron oxide nanoparticles by functional groups such as -NH<sub>2</sub>, -SH etc. encapsulating the iron/iron oxide nanoparticles with other materials, such as humic acid, polyethylenimine, polyrhodanine, MnO<sub>2</sub>, polypyrrole, etc. to make a core-shell structure coating the iron/iron oxide nanoparticles on some porous materials such as graphene oxide, CNTs, and so on.

#### ***Recent advances in supramoleculs***

Poisons always have fascinated humankind. Initially considered as deleterious or hazardous substances, the modern era has witnessed the controlled utilization of dangerous poisons in medicine and cosmetics. Simultaneously, antidotes have become crucial as reversal agents to counteract the effects of a poison, and they are also used today to positively cancel the benefits of a poison after use. Currently, the majority of poisons are composed of

small molecules. This review focuses on recent developments to reverse or prevent toxic effects of poisons by encapsulation in host molecules. Cyclodextrins, cucurbiturils, acyclic cucurbituril derivatives, calixarenes, and pillararenes, have been reported to largely impact the effects of toxic compounds, thus extending the current paradigm of small molecule antidotes by adding a new family of macrocyclic compounds to the current arsenal of antidotes.

1) Supramolecular "trap" of spermine for cancer treatment Polyamines, including spermine (SPM), spermidine (SPD), and putrescine (PUT), exist in a wide range of living organisms and are essential for cell proliferation and differentiation. Therefore, elimination of free polyamines in cancer cells has become a potential approach to induce cancer cell apoptosis to improve cancer treatment.

### 2) *Supramolecular sequestration of cholesterol for treatment of atherosclerosis and NPC disease*

Cholesterol plays a vital role in the progression of atherosclerosis (among several other diseases), since cholesterol is a major component of atherosclerotic plaques and its accumulation and deposition triggers a complex inflammatory response to further promote atherosclerosis. 2-Hydroxypropyl- $\beta$ -cyclodextrin (HP- $\beta$ -CD), an FDA-approved dextrin to enhance the solubility of numerous lipophilic agents, was found to enhance the solubility of cholesterol.

### 3) *Encapsulation of reactive radicals*

One of the main scientific issues of radical chemistry is the reactivity and stability control of organic radicals. Generally, organic radicals can be either thermodynamically stabilized or kinetically stabilized through the effective delocalization of spin density and protection, mainly via covalent approaches. Alternatively, the supramolecular strategy has also been successfully employed to modulate the stability of radicals. Actually, stabilization of reactive species or transient reaction intermediates by supramolecular encapsulation is not surprising. For instance, many studies in the field of zeolites demonstrated that short lived carbon-centered radicals became persistent when they were located inside the zeolite channels. Besides, endohedral fullerenes with nitrogen atoms and metals also highlighted the importance of encapsulation on stabilizing the reactive species. In the field of supramolecular radical.

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## CONCLUSION:

Representative progress in the nanotechnology-based analysis and removal of heavy metals has been summarized in this review. For heavy metal analysis, nanomaterials can be incorporated as different roles of adsorbent, filter membrane, reducing agent, peroxide catalyst, and conjugator. The detection can be based on electrochemical, colorimetric, fluorescent and biosensing technologies.

Various nanomaterials, such as carbon-based, magnetic semiconducting, and noble metal nanoparticles, have been introduced and discussed. For heavy metal removal, magnetic nanoparticles are ideal adsorbents since they can be easily recovered and collected by an external magnet. On the contrary, using other nanomaterials (that with nonmagnetism) to separate targets from complex matrices involves cumbersome procedures, such as filtration or centrifuge, which inevitably increase the separation costs.

Nanocomposite materials exhibit outstanding performance in both analysis and removal of heavy metals. As heavy metal nanosensors, such materials have advantages of wide line range, low detection limit, high sensitivity, and good selectivity. However, since they are in the early stages of development, they currently have defects of fluctuated stability, low practicability in field, high synthetic price, and

Complex synthetic procedures. As heavy metal nanoremovalers, nanocomposites are drawing concerns about the toxicity of some interface materials (including mercury and QDs), especially when they are designed for food, water, and other environmental samples. Current safety standards of nanomaterials are suggested to be reviewed on a relatively frequent basis, in order to keep up with the emerging field.

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